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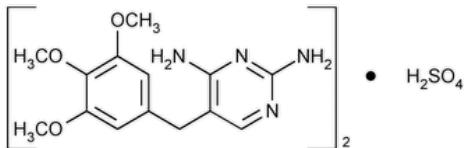
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## Trimethoprim Sulfate



$(\text{C}_{14}\text{H}_{18}\text{N}_4\text{O}_3)_2 \cdot \text{H}_2\text{SO}_4$  678.71

2,4-Pyrimidinediamine, 5-[(3,4,5-trimethoxyphenyl)methyl], sulfate (2:1) (salt).

2,4-Diamino-5-[(3,4,5-trimethoxybenzyl)pyrimidine]-, sulfate (2:1) (salt) CAS RN®: 56585-33-2; UNII: E377MF8EQ8.

» Trimethoprim Sulfate contains not less than 98.5 percent and not more than 101.0 percent of  $(\text{C}_{14}\text{H}_{18}\text{N}_4\text{O}_3)_2 \cdot \text{H}_2\text{SO}_4$ , calculated on the anhydrous basis.

**Packaging and storage**—Preserve in well-closed containers. Store at 25°, excursions permitted between 15° and 30°.

**USP REFERENCE STANDARDS (11)**—

[USP Trimethoprim RS](#)

### Identification—

**Change to read:**

**A:** [▲ Spectroscopic Identification Tests \(197\), Ultraviolet-Visible Spectroscopy: 197U](#) ▲ (CN 1-May-2020) —

**Solution**—Transfer about 100 mg of it, accurately weighed, to a 100-mL volumetric flask, dissolve in 25 mL of alcohol, dilute with 0.1 N sodium hydroxide to volume, and mix.

**Medium**—Dilute the **Solution** quantitatively and stepwise with 0.1 N sodium hydroxide to obtain a solution containing a known concentration of about 20 µg per mL.

Absorptivity, at about 287 nm, calculated on the anhydrous basis, is between 83.0% and 86.4% of [USP Trimethoprim RS](#).

**B:** It responds to the tests for [Sulfate \(191\)](#).

**MELTING RANGE (741):** between 210° and 215°.

**pH (791):** between 7.5 and 8.5, in a solution (0.5 mg per mL).

**WATER DETERMINATION, Method I (921)** : not more than 3.0%.

### Chromatographic purity—

**Adsorbent:** 0.25-mm layer of chromatographic silica gel mixture.

**Diluent**—Prepare a mixture of chloroform and methanol (9:1).

**Test solution**—Transfer about 20 mg of Trimethoprim Sulfate, accurately weighed, to a 10-mL volumetric flask, add 4 mL of glacial acetic acid, and swirl to dissolve. Dilute with **Diluent** to volume, and mix.

**Standard solution**—Dissolve an accurately weighed quantity of [USP Trimethoprim RS](#) in **Diluent**. Dilute an accurately measured volume of this solution quantitatively, and stepwise if necessary, with **Diluent** to obtain a solution having a known concentration of 0.02 mg per mL.

**Application volume:** 10 µL.

**Developing solvent system:** a mixture of chloroform, methanol, and 6 N ammonium hydroxide (95:7.5:1).

**Procedure**—Proceed as directed for *Thin-Layer Chromatography* under [Chromatography \(621\)](#). Spray the plate with a freshly prepared mixture of 1.9 g of ferric chloride in 20 mL of water and 0.5 g of potassium ferricyanide in 10 mL of water. Compare the intensities of any secondary spots observed in the chromatogram of the **Test solution** with that of the principal spot in the chromatogram of the **Standard solution**: no secondary spot in the chromatogram obtained from the **Test solution** is larger or more intense than the principal spot obtained from the **Standard solution** (0.1%); and the sum of the intensities of the secondary spots obtained from the **Test solution** corresponds to not more than 0.5%.

**Assay**—Transfer about 800 mg of Trimethoprim Sulfate, accurately weighed, to a 50-mL conical flask, add about 60 mL of glacial acetic acid, and titrate with 0.1 N perchloric acid VS, determining the endpoint potentiometrically. Perform a blank determination, and make any necessary

correction. Each mL of 0.1 N perchloric acid is equivalent to 67.87 mg of  $(C_{14}H_{18}N_4O_3)_2 \cdot H_2SO_4$ .

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
TRIMETHOPRIM SULFATE	<a href="#">Documentary Standards Support</a>	SM12020 Small Molecules 1
REFERENCE STANDARD SUPPORT	RS Technical Services <a href="mailto:RSTECH@usp.org">RSTECH@usp.org</a>	SM12020 Small Molecules 1

**Chromatographic Database Information:** [Chromatographic Database](#)

**Most Recently Appeared In:**

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