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Sodium Ammonium Phosphate Tetrahydrate

(Microcosmic Salt), $NaNH_4HPO_4 \cdot 4H_2O$ 209.07—Colorless crystals or white granules. Freely soluble in water; insoluble in alcohol. Effloresces in air and loses ammonia.

• **Insoluble Matter and Ammonium Hydroxide Precipitate**

Analysis: Dissolve 10 g in 100 mL of water, add 10 mL of ammonia TS, and heat on a steam bath for 1 hr. If any precipitate is formed, filter, wash well with water, and ignite.

Acceptance criteria: NMT 1 mg (0.01%)

Chloride (Reagent test)**:** One g shows NMT 0.02 mg of Cl (0.002%).

• **Heavy Metals**

Analysis: Dissolve 3 g in 25 mL of water, add 15 mL of 1 N sulfuric acid, then add 10 mL of hydrogen sulfide TS.

Acceptance criteria: Any brown color developed in 1 min is not darker than that of a control containing 3 mL of Standard lead solution TS and 0.5 mL of 1 N sulfuric acid (0.001%).

• **Nitrate**

Analysis: Dissolve 1 g in 10 mL of water, add 0.1 mL of indigo carmine TS, then add, with stirring, 10 mL of sulfuric acid.

Acceptance criteria: Blue color persists for 10 min (about 0.005%).

• **Sulfate** (Reagent test, *Method II*)

Analysis: Dissolve 10 g in 100 mL of water, add 5 mL of hydrochloric acid, and filter if necessary.

Acceptance criteria: NMT 5 mg (0.02%)

Auxiliary Information - Please check for your question in the FAQs before contacting USP.

Topic/Question	Contact	Expert Committee
SODIUM AMMONIUM PHOSPHATE TETRAHYDRATE	Margareth R.C. Marques Principal Scientific Liaison	HDQ Headquarters

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