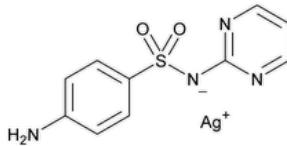


Status: Currently Official on 16-Feb-2025  
Official Date: Official as of 01-May-2020  
Document Type: USP Monographs  
DocId: GUID-C01B43BA-5683-4851-9BBD-AC5F5097C1CB\_2\_en-US  
DOI: [https://doi.org/10.31003/USPNF\\_M78960\\_02\\_01](https://doi.org/10.31003/USPNF_M78960_02_01)  
DOI Ref: m1j6h

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## Silver Sulfadiazine



$C_{10}H_9AgN_4O_2S$  357.14

Benzenesulfonamide, 4-amino-N-2-pyrimidinyl-, monosilver(1+) salt;  
 $N^1$ -2-Pyrimidinylsulfanilamide monosilver(1+) salt CAS RN®: 22199-08-2; UNII: W46JY43EJR.

### DEFINITION

Silver Sulfadiazine contains NLT 98.0% and NMT 102.0% of silver sulfadiazine ( $C_{10}H_9AgN_4O_2S$ ), calculated on the dried basis.

### IDENTIFICATION

**Change to read:**

- A. ▲ [SPECTROSCOPIC IDENTIFICATION TESTS \(197\), Infrared Spectroscopy](#): 197A, 197D, or 197K▲ (CN 1-May-2020)
- B. The retention time of the major peak of the *Sample solution* corresponds to that of the *Standard solution*, as obtained in the Assay.
- C. [IDENTIFICATION TESTS—GENERAL \(191\), Chemical Identification Tests, Silver](#)

**Sample:** 1 g in 15 mL of [ammonium hydroxide](#) and 15 mL of [water](#) in a 50-mL volumetric flask. Dilute with [water](#) to volume, and mix.

**Acceptance criteria:** Meets the requirements

### ASSAY

#### • PROCEDURE

**Solution A:** 0.77 g/L of [ammonium acetate](#) in [water](#). Adjust with [glacial acetic acid](#) to a pH of  $5.2 \pm 0.1$ .

**Solution B:** [Methanol](#)

**Mobile phase:** See [Table 1](#).

Table 1

Time (min)	Solution A (%)	Solution B (%)
0	98	2
4	98	2
16	40	60
20	40	60
22	20	80
32	20	80
34	98	2
40	98	2

**Diluent:** [Acetonitrile](#), [ammonium hydroxide](#), and [water](#) (5:6:90)

**Standard solution:** 0.05 mg/mL of [USP Silver Sulfadiazine RS](#) in *Diluent*. Sonicate, if necessary, to dissolve.

**Sample solution:** 0.05 mg/mL of Silver Sulfadiazine in *Diluent*

#### Chromatographic system

(See [Chromatography \(621\), System Suitability](#).)

**Mode:** LC

**Detector:** UV 275 nm

**Column:** 4.6-mm × 25-cm; 5-μm packing [L1](#)

**Column temperature:** 30°

**Flow rate:** 1 mL/min

**Injection volume:** 10 μL

#### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Tailing factor:** NMT 2.0

**Relative standard deviation:** NMT 1.0%

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of silver sulfadiazine ( $C_{10}H_9AgN_4O_2S$ ) in the portion of Silver Sulfadiazine taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

$r_U$  = peak response from the *Sample solution*

$r_S$  = peak response from the *Standard solution*

$C_S$  = concentration of [USP Silver Sulfadiazine RS](#) in the *Standard solution* (mg/mL)

$C_U$  = concentration of Silver Sulfadiazine in the *Sample solution* (mg/mL)

**Acceptance criteria:** 98.0%–102.0% on the dried basis

#### OTHER COMPONENTS

##### • CONTENT OF SILVER

**Sample solution:** Add 500 mg to 150 mL of [water](#) and 50 mL of [nitric acid](#), and stir for 15 min.

#### Titrimetric system

(See [Titrimetry \(541\)](#).)

**Mode:** Direct titration

**Titrant:** [0.1 N potassium thiocyanate VS](#) or [0.1 N ammonium thiocyanate VS](#)

**Endpoint detection:** Potentiometric

**Analysis:** Titrate the *Sample solution* with *Titrant* using a silver-based indicator electrode and a double-junction reference electrode. Perform a blank determination, and make any necessary correction. Each mL of [0.1 N potassium thiocyanate VS](#) or [0.1 N ammonium thiocyanate VS](#) is equivalent to 10.79 mg of silver.

**Acceptance criteria:** 29.3%–30.5% of silver is found.

#### IMPURITIES

##### • LIMIT OF NITRATE

**Standard solution:** 200 μg/mL of nitrate from [potassium nitrate](#)

**Sample solution:** Add 30.0 mL of [water](#) to 2 g of Silver Sulfadiazine, stir for 20 min, and pass through a suitable, nitrate-free filter.

#### Instrumental conditions

**Mode:** UV-Vis

**Analytical wavelength:** 408 nm

**Blank:** Deionized water

#### Analysis

**Samples:** *Standard solution*, *Sample solution*, and *Blank*

Pipet 3 mL of the *Sample solution* and *Blank* into separate test tubes. Pipet 1 mL of the *Standard solution* and 2 mL of [water](#) into a third test tube. Cool the three test tubes in an ice bath. Slowly add 7.0 mL of cold chromotropic acid solution, prepared by dissolving 50 mg of chromotropic acid in 100 mL of cold sulfuric acid, to each test tube, while swirling, and allow the test tubes to remain in the ice bath

for 3 min after the addition of the chromotropic acid solution. Remove the test tubes from the ice bath, and allow to stand for 30 min.

Concomitantly determine the absorbances of the *Sample solution* and the *Standard solution* with a suitable spectrophotometer, against the *Blank*.

Calculate the percentage of nitrate content in the portion of Silver Sulfadiazine taken:

$$\text{Result} = (A_U/A_S) \times (C_S/C_U) \times 100$$

$A_U$  = absorbance of the *Sample solution*

$A_S$  = absorbance of the *Standard solution*

$C_S$  = concentration of nitrate in the *Standard solution* (μg/mL)

$C_U$  = concentration of Silver Sulfadiazine in the *Sample solution* (μg/mL)

**Acceptance criteria:** NMT 0.1%

• **ORGANIC IMPURITIES**

**Solution A, Solution B, Mobile phase, Diluent, and Chromatographic system:** Proceed as directed in the Assay.

**System suitability solution:** 1.0 mg/mL of [USP Silver Sulfadiazine RS](#) and 0.001 mg/mL each of [USP Sulfanilic Acid RS](#), [USP Sulfanilamide RS](#), and [Sulfaguanidine](#) in *Diluent*

**Standard solution:** 0.001 mg/mL of [USP Silver Sulfadiazine RS](#) in *Diluent*

**Sample solution:** 1.0 mg/mL of Silver Sulfadiazine in *Diluent*

**System suitability**

**Samples:** System suitability solution and Standard solution

[NOTE—The relative retention times for [Sulfaguanidine](#) and sulfanilamide are about 0.65 and 0.71, respectively.]

**Suitability requirements**

**Resolution:** NLT 2.0 between sulfaguanidine and sulfanilamide, System suitability solution

**Relative standard deviation:** NMT 5.0% for the sulfadiazine peak, Standard solution

**Analysis**

**Samples:** Standard solution and Sample solution

Calculate the percentage of each impurity in the portion of Silver Sulfadiazine taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times (1/F) \times 100$$

$r_U$  = peak response of each impurity from the *Sample solution*

$r_S$  = peak response of silver sulfadiazine from the *Standard solution*

$C_S$  = concentration of [USP Silver Sulfadiazine RS](#) in the *Standard solution* (mg/mL)

$C_U$  = concentration of Silver Sulfadiazine in the *Sample solution* (mg/mL)

$F$  = relative response factor (see [Table 2](#))

**Acceptance criteria:** See [Table 2](#). Disregard any impurity peaks less than 0.05%.

**Table 2**

Name	Relative Retention Time	Relative Response Factor	Acceptance Criteria, NMT (%)
Sulfanilic acid	0.27	0.27	0.3
Sulfaguanidine <sup>a,b</sup>	0.65	0.85	—
Sulfanilamide <sup>b</sup>	0.71	1.0	—
2-Aminopyrimidine <sup>c</sup>	0.82	0.48	0.3

Name	Relative Retention Time	Relative Response Factor	Acceptance Criteria, NMT (%)
Sulfadiazine <sup>d</sup>	1.00	—	—
Acetylsulfadiazine <sup>e</sup>	1.14	1.0	0.2
Specified unidentified impurity <sup>f</sup>	1.28	1.0	0.15
Individual unspecified impurity	—	1.0	0.10
Total impurities	—	—	0.5

<sup>a</sup> 4-Amino-N-carbamimidoylbenzenesulfonamide.<sup>b</sup> These are controlled using the limit for individual unspecified impurities.<sup>c</sup> Pyrimidin-2-amine.<sup>d</sup> 4-Amino-N-2-pyrimidinylbenzenesulfonamide.<sup>e</sup> N-{4-[N-(Pyrimidin-2-yl)sulfamoyl]phenyl}acetamide.<sup>f</sup> This compound is a specified unknown impurity. No information is available about the chemical structure or chemical name for this impurity.**SPECIFIC TESTS**

- [Loss on Drying \(731\)](#).

**Analysis:** Dry at 105° for 1 h.**Acceptance criteria:** NMT 0.5%**ADDITIONAL REQUIREMENTS**

- **PACKAGING AND STORAGE:** Preserve in well-closed, light-resistant containers.

- [USP REFERENCE STANDARDS \(11\)](#).

[USP Silver Sulfadiazine RS](#)[USP Sulfanilamide RS](#)4-Aminobenzenesulfonamide.  
C6H8N2O2S 172.20[USP Sulfanilic Acid RS](#)  
4-Aminobenzenesulfonic acid.  
C6H7NO3S 173.19**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

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**Chromatographic Database Information:** [Chromatographic Database](#)**Most Recently Appeared In:**

Pharmacopeial Forum: Volume No.

**Current DocID:** GUID-C01B43BA-5683-4851-9BBD-AC5F5097C1CB\_2\_en-US**DOI:** [https://doi.org/10.31003/USPNF\\_M78960\\_02\\_01](https://doi.org/10.31003/USPNF_M78960_02_01)**DOI ref:** [m1j6h](#)