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## Potassium Chloride

KCl 74.55

Potassium chloride CAS RN®: 7447-40-7.

### DEFINITION

Potassium Chloride contains NLT 99.0% and NMT 100.5% of KCl, calculated on the dried basis.

### IDENTIFICATION

- A. [IDENTIFICATION TESTS—GENERAL, Potassium\(191\)](#).

**Sample solution:** 50 mg/mL

**Acceptance criteria:** Meets the requirements

- B. [IDENTIFICATION TESTS—GENERAL, Chloride\(191\)](#).

**Sample solution:** 50 mg/mL

**Acceptance criteria:** Meets the requirements

### ASSAY

- **PROCEDURE**

**Sample:** 200 mg

**Analysis:** Dissolve the *Sample* in 10 mL of water. Add 10 mL of glacial acetic acid, 75 mL of methanol, and 3 drops of eosin Y TS. Titrate, with shaking, with 0.1 N silver nitrate VS to a pink endpoint. Each mL of 0.1 N silver nitrate is equivalent to 7.455 mg of KCl.

**Acceptance criteria:** 99.0%–100.5% on the dried basis

### IMPURITIES

**Change to read:**

- [▲ ALUMINUM \(206\), Procedure 1 ▲ \(CN 1-Jun-2023\)](#) (where it is labeled as intended for use in hemodialysis): Proceed as directed, using 2.0 g of Potassium Chloride to prepare the *Test preparation*.

**Acceptance criteria:** NMT 1 ppm

- **SODIUM**

**Sample solution:** 50 mg/mL

**Acceptance criteria:** *Sample solution* tested on a platinum wire does not impart a pronounced yellow color to a nonluminous flame.

- **IODIDE**

**Standard stock solution:** 1.64 mg/mL of potassium iodide in water

**Standard solution:** Dilute 1.0 mL of *Standard stock solution* with water to 25 mL. Dilute 2.0 mL of this solution with water to 8 mL. Add 1 mL each of chloroform and diluted hydrochloric acid, then add 2 drops of a chloramine T solution (0.1 in 100), and shake gently.

**Sample solution:** Dissolve 2 g of Potassium Chloride in 8 mL of water. Add 1 mL each of chloroform and diluted hydrochloric acid, then add 2 drops of a chloramine T solution (0.1 in 100), and shake gently.

**Acceptance criteria:** The violet color of the chloroform layer is not darker than that of a concomitantly prepared *Standard solution* (NMT 0.005%).

- **BROMIDE**

**Standard stock solution:** 1.28 mg/mL of sodium bromide in water

**Standard solution:** Dilute 2.0 mL of *Standard stock solution* with water to 8 mL. Add 1 mL each of chloroform and diluted hydrochloric acid, then add 5 drops of a chloramine T solution (1 in 100), and shake gently.

**Sample solution:** Dissolve 2 g of Potassium Chloride in 8 mL of water. Add 1 mL each of chloroform and diluted hydrochloric acid, then add 5 drops of a chloramine T solution (1 in 100), and shake gently.

**Acceptance criteria:** The brown color of the chloroform layer is not darker than that of a concomitantly prepared *Standard solution* (NMT 0.1%).

- **CALCIUM AND MAGNESIUM**

**Sample solution:** 10 mg/mL in water

**Analysis:** To 20 mL of *Sample solution* add 2 mL each of 6 N ammonium hydroxide, ammonium oxalate TS, and dibasic sodium phosphate TS.

**Acceptance criteria:** No turbidity is produced within 5 min.

**SPECIFIC TESTS**

- **ACIDITY OR ALKALINITY:** To a solution of 5.0 g in 50 mL of carbon dioxide-free water add 3 drops of phenolphthalein TS: no pink color is produced. Then add 0.30 mL of 0.020 N sodium hydroxide: a pink color is produced.
- **Loss on Drying (731):** Dry a sample at 105° for 2 h: it loses NMT 1.0% of its weight.

**ADDITIONAL REQUIREMENTS**

- **PACKAGING AND STORAGE:** Preserve in well-closed containers.
- **LABELING:** Where Potassium Chloride is intended for use in hemodialysis, it is so labeled.

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
POTASSIUM CHLORIDE	<a href="#">Documentary Standards Support</a>	SM52020 Small Molecules 5
REFERENCE STANDARD SUPPORT	RS Technical Services <a href="mailto:RSTECH@usp.org">RSTECH@usp.org</a>	SM52020 Small Molecules 5

**Chromatographic Database Information:** [Chromatographic Database](#)

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