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# Pilocarpine Nitrate

$C_{11}H_{16}N_2O_2 \cdot HNO_3$  271.27  
2(3*H*)-Furanone, 3-ethylidihydro-4-[(1-methyl-1*H*-imidazol-5-yl)methyl]-, (3*S*-*cis*)-, mononitrate.

Pilocarpine mononitrate CAS RN®: 148-72-1; UNII: M20T465H6J.  
» Pilocarpine Nitrate contains not less than 98.5 percent and not more than 101.0 percent of  $C_{11}H_{16}N_2O_2 \cdot NO_3$ , calculated on the dried basis.

**Packaging and storage**—Preserve in tight, light-resistant containers.

**USP REFERENCE STANDARDS** (11).—  
[USP Pilocarpine Nitrate RS](#)

**Identification**—

**Change to read:**

**A:** ▲ [Spectroscopic Identification Tests \(197\)](#), [Infrared Spectroscopy: 197K](#) ▲ (CN 1-May-2020) ·

**B:** Mix a solution (1 in 10) with an equal volume of ferrous sulfate TS, and superimpose the mixture upon 5 mL of sulfuric acid contained in a test tube: the zone of contact becomes brown.

**MELTING RANGE (741):** between 171° and 176°, with decomposition, but the range between beginning and end of melting does not exceed 3°.

**SPECIFIC ROTATION (781S):** between +79.5° and +82.5°.

*Test solution:* 20 mg per mL, in water.

**LOSS ON DRYING (731)**—Dry it at 105° for 2 hours: it loses not more than 2.0% of its weight.

**READILY CARBONIZABLE SUBSTANCES (271)**—Dissolve 100 mg in 5 mL of sulfuric acid : the solution has no more color than *Matching Fluid A*.

**Chloride**—To 5 mL of a solution (1 in 50), acidified with nitric acid, add a few drops of silver nitrate TS: no opalescence is produced immediately.

**Other alkaloids**—Dissolve 200 mg in 20 mL of water, and divide the solution into two portions. To one portion add a few drops of 6 N ammonium hydroxide and to the other add a few drops of potassium dichromate TS: no turbidity is produced in either solution.

**Assay**—Dissolve about 600 mg of Pilocarpine Nitrate, accurately weighed, in 30 mL of glacial acetic acid, warming slightly to effect solution. Cool to room temperature, and titrate with 0.1 N perchloric acid VS, determining the endpoint potentiometrically. Perform a blank determination, and make any necessary correction. Each mL of 0.1 N perchloric acid is equivalent to 27.13 mg of  $C_{11}H_{16}N_2O_2 \cdot NO_3$ .

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
PILOCARPINE NITRATE	<a href="#">Documentary Standards Support</a>	SM32020 Small Molecules 3
REFERENCE STANDARD SUPPORT	RS Technical Services <a href="mailto:RSTECH@usp.org">RSTECH@usp.org</a>	SM32020 Small Molecules 3

**Chromatographic Database Information:** [Chromatographic Database](#)

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