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# Phosphoric Acid

$\text{H}_3\text{PO}_4$  98.00  
 Phosphoric acid CAS RN®: 7664-38-2.

## DEFINITION

Phosphoric Acid contains NLT 85.0% and NMT 88.0%, by weight, of  $\text{H}_3\text{PO}_4$ .

[CAUTION—Avoid contact, because Phosphoric Acid rapidly destroys tissues.]

## IDENTIFICATION

- [A. IDENTIFICATION TESTS—GENERAL, Phosphate\(191\)](#): When carefully neutralized with 1 N sodium hydroxide, phenolphthalein TS being used as the indicator, it meets the requirements.

## ASSAY

### PROCEDURE

**Sample:** 1 g

**Blank:** 120 mL of water

### Titrimetric system

(See [Titrimetry \(541\)](#).)

**Mode:** Direct titration

**Titrant:** 1 N sodium hydroxide VS

**Endpoint detection:** Visual

**Analysis:** Place the *Sample* in a tared, glass-stoppered flask, and dilute it with water to 120 mL. Add 0.5 mL of thymolphthalein TS. Titrate with 1 N sodium hydroxide VS to the first appearance of a blue color. Perform a blank determination.

Calculate the percentage of phosphoric acid ( $\text{H}_3\text{PO}_4$ ) in the *Sample* taken:

$$\text{Result} = \{[(V_s - V_b) \times N \times F]/W\} \times 100$$

$V_s$  = volume of *Titrant* consumed by the *Sample* (mL)

$V_b$  = volume of *Titrant* consumed by the *Blank* (mL)

$N$  = actual normality of the *Titrant* (mEq/mL)

$F$  = equivalency factor, 49.00 mg/mEq

$W$  = weight of the *Sample* (mg)

**Acceptance criteria:** 85.0%–88.0% by weight

## IMPURITIES

### LIMIT OF NITRATE

**Sample solution:** Dilute 6 mL of Phosphoric Acid with 14 mL of water.

**Analysis:** Mix 5 mL of the *Sample solution* with about 0.1 mL of indigo carmine TS, then add 5 mL of sulfuric acid.

**Acceptance criteria:** The blue color is not discharged within 1 min.

## SPECIFIC TESTS

### CHLORIDE AND SULFATE, [Sulfate\(221\)](#)

**Sample solution:** Dilute 6 mL of Phosphoric Acid with 90 mL of water.

**Analysis:** To the *Sample solution* add 1 mL of barium chloride TS.

**Acceptance criteria:** No precipitate is formed immediately.

### ALKALI PHOSPHATES

**Sample:** 1 mL

**Analysis:** Transfer the *Sample* to a graduated cylinder, and add 6 mL of ether and 2 mL of alcohol.

**Acceptance criteria:** No turbidity is produced.

• **PHOSPHOROUS OR HYPOPHOSPHOROUS ACID**

**Sample solution:** Dilute 6 mL of Phosphoric Acid with 14 mL of water.

**Analysis:** Gently warm 5 mL of the *Sample solution*, and add 2 mL of silver nitrate TS.

**Acceptance criteria:** The mixture does not become brownish.

**ADDITIONAL REQUIREMENTS**

- **PACKAGING AND STORAGE:** Preserve in tight containers.

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
PHOSPHORIC ACID	<a href="#">Documentary Standards Support</a>	SE2020 Simple Excipients
REFERENCE STANDARD SUPPORT	RS Technical Services <a href="mailto:RSTECH@usp.org">RSTECH@usp.org</a>	SE2020 Simple Excipients

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