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# Nitric Acid

$\text{HNO}_3$  63.01  
Nitric acid CAS RN®: 7697-37-2.

## DEFINITION

Nitric Acid contains NLT 69.0% and NMT 71.0%, by weight, of nitric acid ( $\text{HNO}_3$ ). [CAUTION—Avoid contact, because Nitric Acid rapidly destroys tissues.]

## IDENTIFICATION

- **A. IDENTIFICATION TESTS—GENERAL, Nitrate (191):** Meets the requirements

## ASSAY

### PROCEDURE

**Sample solution:** Weigh 2 mL of Nitric Acid in a glass-stoppered conical flask, and add 25 mL of water. Add methyl red TS.  
**Analysis:** Titrate the *Sample solution* with 1 N sodium hydroxide VS. Each mL of 1 N sodium hydroxide is equivalent to 63.01 mg of  $\text{HNO}_3$  (see [Titrimetry \(541\)](#)).  
**Acceptance criteria:** 69.0%–71.0%

## IMPURITIES

- **RESIDUE ON IGNITION (281).**  
**Sample:** 70 mL (100 g)  
**Analysis:** Place the *Sample* in a tared crucible, add 2 drops of sulfuric acid, and evaporate to dryness. Ignite for 15 min.  
**Acceptance criteria:** NMT 0.5 mg (5 ppm)
- **CHLORIDE AND SULFATE, Chloride (221).**  
**Sample:** 35 mL (50 g)  
**Control:** 35  $\mu\text{L}$  of 0.020 N hydrochloric acid  
**Acceptance criteria:** NMT 0.5 ppm; the *Sample* shows no more chloride than corresponds to the *Control*.
- **CHLORIDE AND SULFATE, Sulfate (221).**  
**Sample:** 28 mL  
**Control:** 40  $\mu\text{L}$  of 0.020 N sulfuric acid in an equal volume of solution containing the quantities of reagents used in the analysis  
**Analysis:** Add 10 mg of sodium carbonate to the *Sample*. Evaporate to dryness, dissolve in a mixture of 4 mL of water and 1 mL of dilute hydrochloric acid (50 mg/mL), and filter if necessary. Wash with two 2-mL portions of water, dilute with water to 10 mL, and add 1 mL of barium chloride TS. Observe 10 min after adding the barium chloride.  
**Acceptance criteria:** 1 ppm; any turbidity produced by the *Sample* is not greater than that produced by the *Control*.

### Change to read:

- **IRON (241), Procedures, Procedure 1** ▲ (CN 1-JUN-2023)  
**Sample:** 35 mL (50 g)  
**Analysis:** Evaporate the *Sample* to dryness, dissolve the residue in 2 mL of hydrochloric acid, and dilute with water to 47 mL.  
**Acceptance criteria:** NMT 0.2  $\mu\text{g/g}$

## SPECIFIC TESTS

### CLARITY AND COLOR OF SOLUTION

**Analysis:** Mix it in its original container, and transfer 10 mL to a 20- × 150-mm test tube. Compare with water in a similar test tube.  
**Acceptance criteria:** The liquids are equally clear and free from suspended matter, and when viewed transversely by transmitted light, exhibit no apparent difference in color.

## ADDITIONAL REQUIREMENTS

- **PACKAGING AND STORAGE:** Preserve in tight containers.

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
NITRIC ACID	<a href="#">Documentary Standards Support</a>	SE2020 Simple Excipients
REFERENCE STANDARD SUPPORT	RS Technical Services <a href="mailto:RSTECH@usp.org">RSTECH@usp.org</a>	SE2020 Simple Excipients

**Chromatographic Database Information:** [Chromatographic Database](#)

**Most Recently Appeared In:**

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