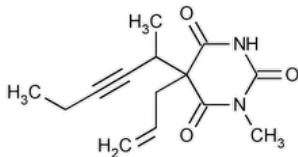


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## Methohexital



$C_{14}H_{18}N_2O_3$  262.30

2,4,6(1*H*,3*H*,5*H*)-Pyrimidinetrione, 1-methyl-5-(1-methyl-2-pentynyl)-5-(2-propenyl)-, ( $\pm$ )-.

( $\pm$ )-5-Allyl-1-methyl-5-(1-methyl-2-pentynyl)barbituric acid CAS RN®: 151-83-7; UNII: E5B8ND5IPE.

» Methohexital contains not less than 98.0 percent and not more than 101.0 percent of  $C_{14}H_{18}N_2O_3$ , calculated on the anhydrous basis.

**Packaging and storage**—Preserve in well-closed containers.

### USP REFERENCE STANDARDS (11)—

[USP Methohexital RS](#)

**Change to read:**

▲ [SPECTROSCOPIC IDENTIFICATION TESTS \(197\)](#), *Infrared Spectroscopy: 197S* ▲ (CN 1-May-2020) —

**Solution:** 1 in 100.

**Medium:** chloroform.

**MELTING RANGE (741):** between 92° and 96°, but the range between beginning and end of melting does not exceed 3°.

**WATER DETERMINATION, Method I (921):** not more than 2.0%.

**CHLORIDE (221):**—Dissolve 200 mg in a mixture of 75 mL of ether and 25 mL of water, agitate, and allow to separate: the water solution shows no more chloride than corresponds to 0.17 mL of 0.010 N hydrochloric acid (0.03%).

### ORDINARY IMPURITIES (466)—

**Test solution:** methanol.

**Standard solution:** methanol.

**Eluant:** a mixture of chloroform and acetone (7:3).

**Visualization**—Expose the plate to chlorine gas for 1 minute, and air-dry the plate at room temperature for 2 minutes. Prepare a solution of 0.5 g of potassium iodide in 50 mL of water, and prepare a solution of 1.5 g of soluble starch in 50 mL of hot water. Mix 10 mL of each solution with 4 mL of alcohol to obtain the *Detection reagent*. [NOTE—The *Detection reagent* so obtained may be used for up to 3 or 4 days.] Spray the plate with the *Detection reagent*.

**Assay**—Dissolve about 100 mg of Methohexital, accurately weighed, in chloroform, and dilute quantitatively and stepwise with chloroform to obtain a solution having a concentration of about 10 mg per mL. Dissolve an accurately weighed quantity of [USP Methohexital RS](#) in chloroform, and dilute quantitatively and stepwise with chloroform to obtain a Standard solution having a known concentration of about 10 mg per mL. Concomitantly determine the absorbances of both solutions in 0.1-mm cells at the wavelength of maximum absorbance at about 5.93  $\mu$ m, with a suitable spectrophotometer, using chloroform as the blank. Calculate the quantity, in mg, of  $C_{14}H_{18}N_2O_3$  in the portion of

Methohexital taken by the formula:

$$10C(A_u/A_s)$$

in which C is the concentration, in mg per mL, of [USP Methohexital RS](#) in the Standard solution; and  $A_u$  and  $A_s$  are the absorbances of the solution of Methohexital and the *Standard solution*, respectively.

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
METHOHEXITAL	<a href="#">Documentary Standards Support</a>	SM52020 Small Molecules 5

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