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# Magnesium Oxide Capsules

## DEFINITION

Magnesium Oxide Capsules contain NLT 90.0% and NMT 110.0% of the labeled amount of magnesium oxide (MgO).

## IDENTIFICATION

- **A. IDENTIFICATION TESTS—GENERAL (191), *Chemical Identification Tests, Magnesium***

**Sample solution:** Transfer the contents of 1 Capsule to a beaker. Add 10 mL of 3 N [hydrochloric acid](#) and 5 drops of [methyl red TS](#), and heat to boiling. Add [6 N ammonium hydroxide](#) until the color of the solution changes to deep yellow, then continue boiling for 2 min, and filter. Use the filtrate.

**Acceptance criteria:** Meet the requirements

**Add the following:**

- ▲ **B.** The retention time of the magnesium peak of the *Sample solution* corresponds to that of the *Standard solution*, as obtained in the *Assay*. ▲ (Official 1-Jun-2021)

## ASSAY

**Change to read:**

### • PROCEDURE

▲ Use water with a resistivity of NLT 18 megohm-cm to prepare the solutions.

**Mobile phase:** 48 mM [methanesulfonic acid](#)

[NOTE—It is recommended to use suitable cation trapping techniques to ensure the *Mobile phase* is free of all cationic impurities.]

**Diluent:** 0.02 N [hydrochloric acid](#)

**System suitability solution:** 33 µg/mL of [USP Magnesium Oxide RS](#) and 5 µg/mL of [USP Calcium Carbonate RS](#) in *Diluent*

**Standard stock solution:** 3.3 mg/mL of [USP Magnesium Oxide RS](#) prepared as follows. Transfer an appropriate portion of [USP Magnesium Oxide RS](#) to a suitable volumetric flask. Add about 20% of the final volume of 6 N [hydrochloric acid](#), and dissolve. Dilute with [water](#) to volume.

**Standard solution:** 33 µg/mL of [USP Magnesium Oxide RS](#) in [water](#) from the *Standard stock solution*

**Sample stock solution:** Nominally 3.3 mg/mL of magnesium oxide prepared as follows. Mix and finely powder the contents of NLT 30 Capsules and transfer an appropriate portion of the powder to a suitable container. Add about 20% of the final volume of 6 N [hydrochloric acid](#). Heat to boiling with constant swirling for 10 min. Allow to cool to room temperature. Transfer the solution quantitatively to a suitable volumetric flask containing 10% of the flask volume of [water](#). Dilute with [water](#) to volume. Pass through a suitable filter of 0.45-µm pore size.

**Sample solution:** Nominally 33 µg/mL of magnesium oxide in [water](#) from the *Sample stock solution*

### Chromatographic system

(See [Chromatography \(621\), System Suitability](#).)

**Mode:** LC

**Detector:** Conductivity with suppression

### Columns

**Guard:** 5-mm × 5-cm; 5.5-µm packing [L84](#)

**Analytical:** 5-mm × 25-cm; 5.5-µm packing [L84](#)

**Column temperature:** 40°

**Flow rate:** 1.0 mL/min

**Injection volume:** 10 µL

**Run time:** NLT 2 times the retention time of magnesium

### System suitability

**Samples:** *System suitability solution* and *Standard solution*

[NOTE—The relative retention times for the magnesium and calcium ions are 1.0 and 1.3, respectively.]

### Suitability requirements

**Resolution:** NLT 3.0 between the magnesium and calcium ions, *System suitability solution*

**Tailing factor:** NMT 2.0, *Standard solution*

**Relative standard deviation:** NMT 2.0%, *Standard solution*

### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of magnesium oxide (MgO) in the portion of Capsules taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times 100$$

$r_U$  = peak response of magnesium from the *Sample solution*

$r_S$  = peak response of magnesium from the *Standard solution*

$C_S$  = concentration of [USP Magnesium Oxide RS](#) in the *Standard solution* (µg/mL)

$C_U$  = nominal concentration of magnesium oxide in the *Sample solution* (µg/mL)▲ (Official 1-Jun-2021)

**Acceptance criteria:** 90.0%–110.0%

**PERFORMANCE TESTS**

• [DISSOLUTION \(711\)](#):

**Medium:** 0.1 N [hydrochloric acid](#); 900 mL

**Apparatus 1:** 100 rpm

**Time:** 45 min

**Analysis:** Using atomic absorption spectrophotometry at a wavelength of 285.2 nm, determine the amount of magnesium oxide (MgO) dissolved, using filtered portions of the solution under test, suitably diluted with *Medium* if necessary, in comparison with a standard solution having a known concentration of magnesium in the same *Medium*.

**Tolerances:** NLT 75% (Q) of the labeled amount of magnesium oxide (MgO) is dissolved.

• [UNIFORMITY OF DOSAGE UNITS \(905\)](#): Meet the requirements

**SPECIFIC TESTS**

• [ACID-NEUTRALIZING CAPACITY \(301\)](#): NLT 5 mEq of acid is consumed by the minimum single dose recommended in the labeling, and NLT 85.0% of the expected mEq value calculated from the results of the Assay is obtained. Each mg of magnesium oxide (MgO) has an expected acid-neutralizing capacity value of 0.0492 mEq.

**ADDITIONAL REQUIREMENTS**

**Change to read:**

• **PACKAGING AND STORAGE:** Preserve in ▲tight▲ (Official 1-Jun-2021) containers.

**Add the following:**

▲• [USP REFERENCE STANDARDS \(11\)](#):

[USP Calcium Carbonate RS](#)

[USP Magnesium Oxide RS](#)▲ (Official 1-Jun-2021)

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
MAGNESIUM OXIDE CAPSULES	<a href="#">Documentary Standards Support</a>	SM32020 Small Molecules 3

**Chromatographic Database Information:** [Chromatographic Database](#)

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