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## Magaldrate and Simethicone Oral Suspension

» Magaldrate and Simethicone Oral Suspension contains not less than 90.0 percent and not more than 110.0 percent of the labeled amount of magaldrate  $[\text{Al}_5\text{Mg}_{10}(\text{OH})_{31}(\text{SO}_4)_2]$ , and an amount of polydimethylsiloxane  $[-(\text{CH}_3)_2\text{SiO}-]^n$  that is not less than 85.0 percent and not more than 115.0 percent of the labeled amount of simethicone.

**Packaging and storage**—Preserve in tight containers, and keep from freezing.

### USP REFERENCE STANDARDS (11)—

[USP Magaldrate RS](#)

[USP Polydimethylsiloxane RS](#)

### **Identification**—

**A:** Dissolve an amount of Oral Suspension, equivalent to about 800 mg of magaldrate, in 20 mL of 3 N hydrochloric acid, dilute with water to about 50 mL, add 3 drops of methyl red TS, and proceed as directed in *Identification* test A under [Magaldrate](#), beginning with "and heat to boiling."

**B:** Wash the precipitate obtained in *Identification* test A with hot ammonium chloride solution (1 in 50), and dissolve the precipitate in hydrochloric acid. Divide this solution into two portions: the dropwise addition of 6 N ammonium hydroxide to one portion yields a gelatinous white precipitate, which does not dissolve in an excess of 6 N ammonium hydroxide. The dropwise addition of 1 N sodium hydroxide to the other portion yields a gelatinous white precipitate, which dissolves in an excess of 1 N sodium hydroxide, leaving some turbidity.

**C:** Transfer an amount of Oral Suspension, equivalent to about 1 g of magaldrate, to a 100-mL centrifuge tube. Add about 60 mL of water, insert the cap, and shake for 3 minutes. Centrifuge the suspension, and discard the supernatant. Repeat the washing of the residue with three 60-mL portions of water. Transfer the residue to a 250-mL beaker, and heat on a steam bath to dryness: the X-ray diffraction pattern (see [X-Ray Powder Diffraction \(941\)](#)), in the d-spacings region below 2.57 angstrom units, of the residue so obtained conforms to that of [USP Magaldrate RS](#).

**D:** The IR absorption spectrum, in the 7- to 15- $\mu\text{m}$  region, determined in a 0.1-mm cell, of the *Assay preparation* prepared as directed in the [Assay for polydimethylsiloxane](#) exhibits maxima only at the same wavelengths as that of the *Standard preparation* prepared as directed in the [Assay for polydimethylsiloxane](#).

**MICROBIAL ENUMERATION TESTS (61) and TESTS FOR SPECIFIED MICROORGANISMS (62)**—Its total aerobic microbial count does not exceed 100 cfu per mL, and it meets the requirements of the test for absence of *Escherichia coli*.

**ACID-NEUTRALIZING CAPACITY (301)**—The acid consumed by the minimum single dose recommended in the labeling is not less than 5 mEq, and not less than the number of mEq calculated by the formula:

$$0.8(0.0282M)$$

in which 0.0282 is the theoretical acid-neutralizing capacity, in mEq per mg, of magaldrate; and  $M$  is the quantity, in mg, of the labeled amount of magaldrate.

### **Magnesium hydroxide content**—

*Test preparation*—Transfer an accurately measured quantity of Oral Suspension, equivalent to about 1 g of magaldrate, to a 100-mL volumetric flask, add 30 mL of dilute hydrochloric acid (1 in 10), shake to dissolve, dilute with water to volume, and mix.

*Procedure*—Transfer 10.0 mL of *Test preparation* to a 400-mL beaker, and proceed as directed in the test for [Magnesium hydroxide content](#) under [Magaldrate](#), beginning with "and dilute with water to about 200 mL." Not less than 492 mg and not more than 666 mg of magnesium hydroxide  $[\text{Mg}(\text{OH})_2]$  per g of the labeled amount of magaldrate is found.

### **Aluminum hydroxide content**—

*Edetate disodium titrant*—Prepare and standardize as directed in the [Assay](#) under [Ammonium Alum](#).

*Test preparation*—Prepare as directed in the test for [Magnesium hydroxide content](#).

*Procedure*—Transfer 10.0 mL of *Test preparation* and 20 mL of water to a 250-mL beaker, and proceed as directed for [Procedure](#) in the test for [Aluminum hydroxide content](#) under [Magaldrate](#), beginning with "Add, with stirring, 25.0 mL of *Edetate disodium titrant*." Not less than 321 mg and not more than 459 mg of aluminum hydroxide  $[\text{Al}(\text{OH})_3]$  per g of the labeled amount of magaldrate is found.

**Other requirements**—Evaporate a volume of Oral Suspension, equivalent to about 5 g of magaldrate, on a steam bath to dryness: the residue so obtained meets the requirements of the tests for [Arsenic](#) under [Magaldrate](#).

### **Change to read:**

**Assay for magaldrate**—Transfer an accurately measured quantity of Oral Suspension, equivalent to about 3 g of magaldrate, to a beaker. Add 100.0 mL of 1 N hydrochloric acid VS, and mix, using a magnetic stirrer to achieve dissolution. Titrate the excess acid with 1 N sodium

hydroxide VS to a pH of 3.0, determined potentiometrically. Perform a blank determination (see [▲ Titrimeetry \(541\)](#)▲ (CN 1-Aug-2024)). Each mL of 1 N hydrochloric acid is equivalent to 35.40 mg of magaldrate  $[\text{Al}_5\text{Mg}_{10}(\text{OH})_{31}(\text{SO}_4)_2]$ .

**Assay for polydimethylsiloxane**—Transfer an accurately measured quantity of Oral Suspension, equivalent to about 250 mg of simethicone, to a 200-mL centrifuge bottle. Add an equal volume of hydrochloric acid, swirl to dissolve the Oral Suspension, add 25.0 mL of hexanes, and immediately close the bottle securely with a cap having an inert liner. Shake the bottle for 30 minutes, and centrifuge the mixture until a clear supernatant layer is obtained (*Assay preparation*). Prepare a *Standard preparation* of [USP Polydimethylsiloxane RS](#) in hexanes having a known concentration of about 10 mg per mL. Concomitantly determine the absorbances of the *Assay preparation* and the *Standard preparation* in 0.1-mm cells at the wavelength of maximum absorbance at about 7.9  $\mu\text{m}$  and at the wavelengths of minimum absorbance at about 7.5  $\mu\text{m}$  and 8.3  $\mu\text{m}$ , with a suitable IR spectrophotometer, using hexanes as the blank. Draw a linear baseline between the two minima, and determine the absorbances for the *Standard preparation* and the *Assay preparation* with respect to the baseline, making any necessary correction for the blank. Calculate the quantity, in mg, of  $[-(\text{CH}_3)_2\text{SiO}-]^n$  in the portion of Oral Suspension taken by the formula:

$$25C(A_u/A_s)$$

in which C is the concentration, in mg per mL, of [USP Polydimethylsiloxane RS](#) in the *Standard preparation*; and  $A_u$  and  $A_s$  are the absorbances of the *Assay preparation* and the *Standard preparation*, respectively.

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
MAGALDRATE AND SIMETHICONE ORAL SUSPENSION	<a href="#">Documentary Standards Support</a>	SM32020 Small Molecules 3

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