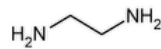


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# Ethylenediamine



$C_2H_8N_2$  60.10  
1,2-Ethanediamine.

Ethylenediamine CAS RN®: 107-15-3; UNII: 60V9STC53F.

» Ethylenediamine contains not less than 98.0 percent and not more than 100.5 percent, by weight, of  $C_2H_8N_2$ .

[**CAUTION**—Use care in handling Ethylenediamine because of its caustic nature and the irritating properties of its vapor.]

[**NOTE**—Ethylenediamine is strongly alkaline and may readily absorb carbon dioxide from the air to form a nonvolatile carbonate. Protect Ethylenediamine against undue exposure to the atmosphere.]

**Packaging and storage**—Preserve in well-filled, tight, glass containers.

**Identification**—To 2 mL of cupric sulfate solution (1 in 100) add 3 drops of a solution of Ethylenediamine (1 in 6), and shake: a purplish blue color is produced.

**Assay**—Weigh accurately about 1 mL of Ethylenediamine in a tared, glass-stoppered flask containing about 25 mL of water. Dilute with water to about 75 mL, add a mixed indicator of bromocresol green TS and methyl red TS (5 in 6), mix, and titrate with 1 N hydrochloric acid VS. Perform a blank determination, and make any necessary correction. Each mL of 1 N hydrochloric acid is equivalent to 30.05 mg of  $C_2H_8N_2$ .

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
ETHYLENEDIAMINE	<a href="#">Documentary Standards Support</a>	SM32020 Small Molecules 3

**Chromatographic Database Information:** [Chromatographic Database](#)

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