Status: Currently Official on 14-Feb-2025
Official Date: Official as of 01-Nov-2021
Document Type: USP Monographs
DocId: GUID-6C0850BF-8919-407C-9D7A-7F62E9161C6A_5_en-US
DOI: https://doi.org/10.31003/USPNF_M27754_05_01
DOI Ref: 01008

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Divalproex Sodium Delayed-Release Capsules

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DEFINITION

Divalproex Sodium Delayed-Release Capsules contain an amount of divalproex sodium equivalent to NLT 90.0% and NMT 110.0% of the labeled amount of valproic acid ($C_8H_{16}O_9$).

IDENTIFICATION

• A. Spectroscopic Identification Tests (197), Infrared Spectroscopy: 197K

Diluent: Acetonitrile and water (1:1)

Standard: Prepare as directed in 197F using USP Valproic Acid RS.

Sample: Dissolve the contents of 20 Capsules in 30 mL of *Diluent* in a 50-mL volumetric flask. Sonicate for 30 min to dissolve. Dilute with *Diluent* to volume. Centrifuge the solution at 3000 rpm for about 20 min. Pipet 20 mL of the supernatant into a separatory funnel. Extract with 50 mL of <u>n-hexane</u>. Collect the <u>n-hexane</u> layer and evaporate the solvent. Cast 1 mg of the liquid obtained after evaporation to sodium chloride (NaCl) windows.

• B. The retention time of the major peak of the Sample solution corresponds to that of the Standard solution, as obtained in the Assay.

ASSAY

• Procedure

Buffer: 6.8 g/L of monobasic potassium phosphate. Adjust with phosphoric acid to a pH of 3.0.

Mobile phase: Acetonitrile and Buffer (2:3)

Diluent: Acetonitrile and water (1:1)

Standard solution: Transfer a suitable amount of <u>USP Valproic Acid RS</u> to a suitable volumetric flask to obtain a solution having a final concentration of 2.5 mg/mL of valproic acid. Add 40% of the flask volume of *Diluent*. Sonicate for 5 min and add 20% of the flask volume of <u>0.1 N hydrochloric acid</u>. Dilute with *Diluent* to volume.

Sample solution: Transfer an amount of contents (from NLT 20 Capsules) to a suitable volumetric flask to obtain a nominal concentration of 2.5 mg/mL of valproic acid. Dissolve in 20% of the flask volume of 0.1 N hydrochloric acid and sonicate for 5 min. Add 60% of the flask volume of *Diluent* and sonicate for an additional 25 min. Dilute with *Diluent* to volume. Centrifuge at 4000 rpm for 10 min and use the clear supernatant.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 215 nm

Column: 4.6-mm × 15-cm; 5-µm packing L1

Flow rate: 1.8 mL/min Injection volume: 20 μL System suitability

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Sample: Standard solution
Suitability requirements

Tailing factor: NMT 2.0 for valproic acid

Relative standard deviation: NMT 2.0% for valproic acid

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Samples: Standard solution and Sample solution

Calculate the percentage of the labeled amount of valproic acid $(C_8H_{16}O_2)$ in the portion of Capsules taken:

Result =
$$(r_{I}/r_{S}) \times (C_{S}/C_{I}) \times 100$$

 r_{ij} = peak response from the Sample solution

 $r_{_{\rm S}}$ = peak response from the Standard solution

C_s = concentration of <u>USP Valproic Acid RS</u> in the Standard solution (mg/mL)

 $C_{_{II}}$ = nominal concentration of valproic acid in the Sample solution (mg/mL)

Acceptance criteria: 90.0%-110.0%

PERFORMANCE TESTS

Change to read:

• **D**ISSOLUTION (711)

Test 1

Medium: Phosphate buffer, pH 7.5 (6.8 g/L of monobasic potassium phosphate and 1.64 g/L of sodium hydroxide in water; adjusted with

0.08 N hydrochloric acid TS to a pH of 7.5); 500 mL, degassed

Apparatus 2: 50 rpm, with sinkers

Times: 2, 4, and 6 h

Buffer and Mobile phase: Prepare as directed in the Assay.

Standard stock solution: 1.6 mg/mL of USP Valproic Acid RS in Mobile phase

Standard solution: 0.26 mg/mL of valproic acid from the Standard stock solution and Medium

Sample solution: Pass a portion of the solution under test through a suitable filter of 0.45-µm pore size. Replace the volume withdrawn with

an equal volume of Medium previously heated at $37.0 \pm 0.5^{\circ}$.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 210 nm

Column: 4.6-mm × 15-cm; 5-µm packing L1

Flow rate: 1.8 mL/minInjection volume: $40 \text{ }\mu\text{L}$ System suitability

Sample: Standard solution **Suitability requirements**

Tailing factor: NMT 2.0 for valproic acid

Relative standard deviation: NMT 2.0% for valproic acid

Analysis

Samples: Standard solution and Sample solution

Calculate the concentration (C_i) of valproic acid ($C_8H_{16}O_2$) in the sample withdrawn from the vessel at each time point (i):

Result_i =
$$(r_{ij}/r_{s}) \times C_{s}$$

 r_{ij} = peak response from the Sample solution

r_s = peak response from the Standard solution

 C_{S} = concentration of the Standard solution (mg/mL)

Calculate the percentage of the labeled amount of valproic acid $(C_8H_{16}O_2)$ dissolved at each time point (i):

Result₁ =
$$C_1 \times V \times (1/L) \times 100$$

Result₂ =
$$[(C_2 \times V) + (C_1 \times V_S)] \times (1/L) \times 100$$

Result₃ =
$$\{(C_3 \times V) + [(C_2 + C_1) \times V_S]\} \times (1/L) \times 100$$

C, = concentration of valproic acid in the portion of sample withdrawn at the specified time point (mg/mL)

V = volume of Medium, 500 mL

L = label claim (mg/Capsule)

V_s = volume of the Sample solution withdrawn at each time point and replaced with Medium (mL)

Tolerances: See Table 1.

Table 1

USP-NF Divalproex Sodium Delayed-Release Capsules

https://trungtamthuoc.com/ Amount **Point** Dissolved Time (i) (h) (%) 1 2 15 - 402 4 70-90

The percentage of the labeled amount of valproic acid (C₈H₁₆O₂) dissolved at each time point conforms to <u>Dissolution (711)</u>, <u>Acceptance</u>

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Test 2: If the product complies with this test, the labeling indicates that the product meets USP Dissolution Test 2.

Procedure A

Medium: 0.05 M phosphate buffer, pH 7.5 (6.8 g/L of monobasic potassium phosphate and 1.64 g/L of sodium hydroxide in water; adjusted with 2 N sodium hydroxide to a pH of 7.5); 500 mL

Apparatus 2: 50 rpm, contents of the Capsule

3

Time: 15 min

Standard solution A: 0.036 mg/mL of USP Valproic Acid RS in Medium. A volume of acetonitrile not exceeding 10% of the total volume may be used to dissolve the valproic acid.

Sample solution A: Pass a portion of the solution under test through a suitable filter of 0.45-µm pore size.

Procedure B

Medium: 0.05 M phosphate buffer, pH 7.5 (6.8 g/L of monobasic potassium phosphate and 1.64 g/L of sodium hydroxide in water; adjusted with 2 N sodium hydroxide to a pH of 7.5); 900 mL

Apparatus 2: 50 rpm, with wire helix sinkers

Time: 4 h

Buffer A: 0.5 g/L of citric acid and 0.4 g/L of dibasic sodium phosphate in water

Buffer B: 6.8 g/L of monobasic potassium phosphate and 1.7 g/L of sodium hydroxide in water; adjusted with phosphoric acid to a pH of

Mobile phase: Acetonitrile, Buffer A, and Buffer B (30:35:35); adjusted with phosphoric acid to a pH of 3.0

Standard solution B: 0.13 mg/mL of USP Valproic Acid RS in Medium. A volume of acetonitrile not exceeding 10% of the total volume may be used to dissolve the valproic acid.

Sample solution B: Pass a portion of the solution under test through a suitable filter of 0.45-µm pore size.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 210 nm

Column: 3.9-mm × 15-cm; 4-µm packing L11

Flow rate: 1.2 mL/min

Injection volume: 200 µL for Standard solution A and Sample solution A; 50 µL for Standard solution B and Sample solution B

System suitability

Sample: Standard solution B **Suitability requirements**

Tailing factor: NMT 2.0 for valproic acid

Relative standard deviation: NMT 2.0% for valproic acid

Analysis

Samples: Standard solution A, Sample solution A, Standard solution B, and Sample solution B

Calculate the percentage of the labeled amount of valproic acid $(C_0H_{16}O_2)$ dissolved at each time point:

Result =
$$(r_U/r_S) \times (C_S/L) \times V \times 100$$

= peak response from Sample solution A or Sample solution B

= peak response from Standard solution A or Standard solution B

= concentration of Standard solution A or Standard solution B (mg/mL)

= label claim (mg/Capsule)

= volume of Medium; 500 mL for Sample solution A, 900 mL for Sample solution B

Tolerances: NMT 20% of the labeled amount of valproic acid $(C_8H_{16}O_2)$ is dissolved in 15 min (Sample solution A); NLT 80% (Q) of the labeled amount of valproic acid $(C_gH_{16}O_2)$ is dissolved in 4 h (Sample solution B). The percentage of the labeled amount of valproic acid

NLT 85

 $(C_g H_{16} O_2)$ dissolved at 4 h conforms to <u>Dissolution (711), Acceptance Table 1</u>.

Test 3: If the product complies with this test, the labeling indicates that the product meets USP Dissolution Test 3.

Medium

Acid stage medium: 0.08 N hydrochloric acid TS; 900 mL

Buffer stage medium: Phosphate buffer, pH 7.5 (6.8 g/L of monobasic potassium phosphate and 1.6 g/L of sodium hydroxide in water, prepared as follows. Transfer suitable quantities of monobasic potassium phosphate and sodium hydroxide to a suitable volumetric flask. Dissolve in 83% of the flask volume of water and adjust with 0.1 N hydrochloric acid, if necessary, to a pH of 7.5. Dilute the resulting solution with water to volume.); 900 mL

Times

Acid stage: 2 h Buffer stage: 4 h

Apparatus 2: 50 rpm, with sinkers

Buffer: 0.25 g/L of <u>citric acid</u>, 0.2 g/L of <u>anhydrous dibasic sodium phosphate</u>, 3.4 g/L of <u>monobasic potassium phosphate</u>, and 0.85 g/L of <u>sodium hydroxide</u> in water

Mobile phase: Acetonitrile and Buffer (45:55); mixed, degassed, and adjusted with phosphoric acid to a pH of 2.5

Standard solution: 0.14 mg/mL of <u>USP Valproic Acid RS</u> prepared as follows. Transfer a portion of <u>USP Valproic Acid RS</u> to a suitable volumetric flask. Dissolve in <u>methanol</u> using 5.0% of the final volume. Dilute with *Buffer stage medium* to final volume and mix.

Sample solutions

Acid stage sample solution: Pass a portion of the solution under test through a suitable filter of 0.45-µm pore size, discarding the first 3 mL of filtrate.

Buffer stage sample solution: Pass a portion of the solution under test through a suitable filter of 0.45-µm pore size, discarding the first 3 mL of filtrate.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 210 nm

Column: 3.9-mm × 15-cm; 4-µm packing L11

Flow rate: 1 mL/min Injection volume: 50 μL System suitability

Sample: Standard solution Suitability requirements Tailing factor: NMT 2.0

Relative standard deviation: NMT 2.0%

Analysis

Samples: Standard solution, Acid stage sample solution, and Buffer stage sample solution

Calculate the percentage of the labeled amount of valproic acid (C_oH₁₆O₂) dissolved at each time point:

Result =
$$(r_{I}/r_{S}) \times (C_{S}/L) \times V \times 100$$

 r_{ij} = peak response from the Acid stage sample solution or the Buffer stage sample solution

 $r_{\rm s}$ = peak response from the Standard solution

C_s = concentration of the Standard solution (mg/mL)

L = label claim (mg/Capsule)

V = volume of the Acid stage medium or the Buffer stage medium, 900 mL

Tolerances: The requirements for the *Acid stage* and the *Buffer stage* must be met.

Acid stage: NMT 30% (Q) of the labeled amount of valproic acid ($C_8H_{16}O_2$) is dissolved in 2 h (*Acid stage sample solution*). The percentage of the labeled amount of valproic acid ($C_9H_{16}O_2$) dissolved at 2 h conforms to <u>Table 2</u>.

Table 2

Level	Number Tested	Criteria
A_{τ}	6	No individual value exceeds Q dissolved.
A_2	6	Average of the 12 units $(A_1 + A_2)$ is NMT Q dissolved; and no individual unit is greater

Level	Number Tested	Criteria
		than Q + 15% dissolved.
A_3	12	Average of the 24 units $(A_1 + A_2 + A_3)$ is NMT Q dissolved; and no individual unit is greater than $Q + 15\%$ dissolved.

Buffer stage: NLT 80% (Q) of the labeled amount of valproic acid ($C_8H_{16}O_2$) is dissolved in 4 h (*Buffer stage sample solution*). The percentage of the labeled amount of valproic acid ($C_8H_{16}O_2$) dissolved at 4 h conforms to <u>Dissolution (711), Acceptance Table 2</u>.

Test 4: If the product complies with this test, the labeling indicates that the product meets USP Dissolution Test 4.

Medium: 0.05 M phosphate buffer, pH 7.5 (6.8 g/L of monobasic potassium phosphate in water; adjusted with 2 N sodium hydroxide to a pH of 7.5); 500 mL

Apparatus 2: 50 rpm **Times:** 2, 4, and 8 h

Buffer A: 0.5 g/L of citric acid and 4 g/L of dibasic sodium phosphate in water

Buffer B: 6.8 g/L of monobasic potassium phosphate and 1.7 g/L of sodium hydroxide in water; adjusted with phosphoric acid to a pH of

7.4

Mobile phase: Acetonitrile, Buffer A, and Buffer B (30:35:35); adjusted with phosphoric acid to a pH of 3.0

Standard solution: 0.25 mg/mL of USP Valproic Acid RS in Medium

Sample solution: Pass a portion of the solution under test through a suitable filter of 0.45-µm pore size.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 210 nm

Column: 3.9-mm × 15-cm; 4-µm packing L11

Column temperature: 30° Flow rate: 1.2 mL/min Injection volume: 50 µL

Run time: NLT 1.5 times the retention time of valproic acid

System suitability

Sample: Standard solution **Suitability requirements**

Tailing factor: NMT 2.0 for valproic acid

Relative standard deviation: NMT 2.0% for valproic acid

Analysis

Samples: Standard solution and Sample solution

Calculate the concentration (C_i) of valproic acid ($C_8H_{16}O_2$) in the sample withdrawn from the vessel at each time point (i):

Result_i =
$$(r_{ij}/r_{s}) \times C_{s}$$

 r_{ij} = peak response from the Sample solution

 $r_{\rm s}$ = peak response from the Standard solution

 C_s = concentration of the Standard solution (mg/mL)

Calculate the percentage of the labeled amount of valproic acid (C₈H₁₆O₂) dissolved at each time point (i):

$$Result_1 = C_1 \times V \times (1/L) \times 100$$

Result₂ = {
$$[C_2 \times (V - V_2)] + (C_1 \times V_2)$$
} × (1/L) × 100

Result₃ =
$$({C_3 \times [V - (2 \times V_S)]} + [(C_2 + C_1) \times V_S]) \times (1/L) \times 100$$

C, = concentration of valproic acid in the portion of sample withdrawn at the specified time point (mg/mL)

V = volume of Medium, 500 mL

L = label claim (mg/Capsule)

V_s = volume of the Sample solution withdrawn at each time point (mL)

https://trungtamthuoc.com/ Tolerances: See <u>Table 3</u>.

Table 3

Time Point (i)	Time (h)	Amount Dissolved (NLT %)
1	2	60
2	4	70
3	8	80

The percentage of the labeled amount of valproic acid ($C_8H_{16}O_2$) dissolved at each time point conforms to <u>Dissolution (711), Acceptance</u>

<u>Table 4.</u>

Test 5: If the product complies with this test, the labeling indicates that the product meets USP Dissolution Test 5.

Medium: 0.05 M phosphate buffer, pH 7.5 (6.8 g/L of monobasic potassium phosphate and 1.64 g/L of sodium hydroxide in water; adjusted with 2 N sodium hydroxide to a pH of 7.5); 900 mL, deaerated

Apparatus 2: 50 rpm with suitable sinkers

Times: 1 and 4 h

Buffer A: 0.5 g/L of citric acid and 0.4 g/L of dibasic sodium phosphate in water

Buffer B: 6.8 g/L of monobasic potassium phosphate and 1.7 g/L of sodium hydroxide in water; adjusted with phosphoric acid to a pH of 7.4

Mobile phase: Acetonitrile, Buffer A, and Buffer B (30:35:35); adjusted with phosphoric acid to a pH of 3.0 **Standard solution:** (L/900) mg/mL of USP Valproic Acid RS in Medium, where L is the label claim in mg/Capsule

 $\textbf{Sample solution:} \ \ \text{Pass a portion of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under test through a suitable filter of 0.45-$\mu m pore size, discarding the first 2-3 mL of the solution under the solution under$

filtrate.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 210 nm

Column: 3.9-mm × 15-cm; 4-µm packing L11

Flow rate: 1.2 mL/min Injection volume: 50 µL

Run time: NLT 1.5 times the retention time of valproic acid

System suitability

Sample: Standard solution
Suitability requirements

Tailing factor: NMT 2.0 for valproic acid

Relative standard deviation: NMT 2.0% for valproic acid

Analysis

Samples: Standard solution and Sample solution

Calculate the concentration (C_i) of valproic acid $(C_8H_{16}O_2)$ in the sample withdrawn from the vessel at each time point (i):

Result_i =
$$(r_{ij}/r_{s}) \times C_{s}$$

 r_{ij} = peak response from the Sample solution

 $r_{\rm s}$ = peak response from the Standard solution

 C_{\circ} = concentration of the Standard solution (mg/mL)

Calculate the percentage of the labeled amount of valproic acid $(C_8H_{16}O_2)$ dissolved at each time point (i):

Result₁ =
$$C_1 \times V \times (1/L) \times 100$$

Result₂ = {
$$[C_2 \times (V - V_S)] + (C_1 \times V_S)$$
} × (1/L) × 100

C; = concentration of valproic acid in the portion of sample withdrawn at the specified time point (mg/mL)

V = volume of Medium, 900 mL

L = label claim (mg/Capsule)

V_s = volume of the Sample solution withdrawn at each time point (mL)

Tolerances: See <u>Table 4</u>.

Table 4

Time Point (i)	Time (h)	Amount Dissolved (%)
1	1	NMT 25
2	4	NLT 80

The percentage of the labeled amount of valproic acid ($C_8H_{16}O_2$) dissolved at each time point conforms to <u>Dissolution (711), Acceptance</u>

<u>Table 2.</u>

▲Test 6: If the product complies with this test, the labeling indicates that the product meets USP Dissolution Test 6.

Medium: 0.05 M phosphate buffer pH 7.5 (dissolve 6.8 g of monobasic potassium phosphate and 1.64 g of sodium hydroxide in each liter of water; adjust with 2 N sodium hydroxide to a pH of 7.5); 900 mL

Apparatus 2: 50 rpm, with wire helix sinkers

Times: 0.5, 1.5, and 4 h

Buffer A: Dissolve 0.5 g of citric acid and 0.4 g of dibasic sodium phosphate in each liter of water.

Buffer B: Dissolve 6.8 g of monobasic potassium phosphate and 1.7 g of sodium hydroxide in each liter of water; adjust with phosphoric acid to a pH of 7.4.

Mobile phase: Acetonitrile, Buffer A, and Buffer B (30:35:35). Adjust with phosphoric acid to a pH of 3.0.

Standard stock solution: 1.62 mg/mL of <u>USP Valproic Acid RS</u> in *Medium* prepared as follows. Transfer an appropriate amount of <u>USP Valproic Acid RS</u> into a suitable volumetric flask. Add 10% of flask volume of <u>acetonitrile</u> and sonicate to dissolve. Add about 60% of flask volume of *Medium*. Allow the solution to reach room temperature. Dilute with *Medium* to volume.

Standard solution: 0.13 mg/mL of USP Valproic Acid RS in Medium

Sample solution: At the specified *Times*, withdraw and pass a portion of the solution under test through a suitable filter of 0.45-µm pore size.

Chromatographic system

(See Chromatography (621), System Suitability.)

Mode: LC

Detector: UV 210 nm

Column: 3.9-mm × 15-cm; 4-µm packing L11

Flow rate: 1.2 mL/min Injection volume: $50 \mu L$

Run time: NLT 2 times the retention time of valproic acid

System suitability

Sample: Standard solution
Suitability requirements
Tailing factor: NMT 2.0

Relative standard deviation: NMT 2.0%

Analysis

Samples: Standard solution and Sample solution

Calculate the concentration (C_i) of valproic acid $(C_8H_{16}O_2)$ in the sample withdrawn from the vessel at each time point (i):

Result_i =
$$(r_U/r_S) \times C_S$$

 $r_{_U}$ = peak response of valproic acid from the Sample solution

 $r_{\rm s}$ = peak response of valproic acid from the Standard solution

 C_S = concentration of <u>USP Valproic Acid RS</u> in the Standard solution (mg/mL)

Calculate the percentage of the labeled amount of valproic acid ($C_8H_{16}O_2$) dissolved at each time point (i):

$$Result_1 = C_1 \times V \times (1/L) \times 100$$

Result₂ =
$$[(C_2 \times V) + (C_1 \times V_2)] \times (1/L) \times 100$$

Result₂ =
$$\{(C_2 \times V) + [(C_2 + C_1) \times V_2]\} \times (1/L) \times 100$$

 C_i = concentration of valproic acid in the portion of sample withdrawn at time point i (mg/mL)

/ = volume of *Medium*, 900 mL

L = label claim (mg/Capsule)

V_s = volume of the Sample solution withdrawn at each time point and replaced with Medium (mL)

Tolerances: See <u>Table 5</u>.

Table 5

Time Point (i)	Time (h)	Amount Dissolved (%)
1	0.5	NMT 20
2	1.5	NMT 65
3	4.0	NLT 80 (Q)

The percentage of the labeled amount of valproic acid $(C_aH_{1a}O_2)$ dissolved at the times specified conforms to <u>Dissolution (711)</u>.

Acceptance Table 4. ▲ (RB 1-Nov-2021)

• **UNIFORMITY OF DOSAGE UNITS (905):** Meet the requirements

ADDITIONAL REQUIREMENTS

- PACKAGING AND STORAGE: Preserve in tight, light-resistant containers at controlled room temperature.
- LABELING: Divalproex Sodium Delayed-Release Capsules may be swallowed whole or may be administered by carefully opening the Capsule and sprinkling the entire contents on a small amount of soft food. This drug/food mixture should be swallowed immediately and not chewed. It should not be stored for future use. When more than one *Dissolution* test is given, the labeling states the *Dissolution* test used only if *Test 1* is not used.
- USP REFERENCE STANDARDS (11)

 USP Valproic Acid RS

Auxiliary Information - Please check for your question in the FAOs before contacting USP.

Topic/Question	Contact	Expert Committee
DIVALPROEX SODIUM DELAYED-RELEASE CAPSULES	<u>Documentary Standards Support</u>	SM42020 Small Molecules 4

Chromatographic Database Information: Chromatographic Database

Most Recently Appeared In:

Pharmacopeial Forum: Volume No. PF 36(2)

Current DocID: GUID-6C0850BF-8919-407C-9D7A-7F62E9161C6A_5_en-US

DOI: https://doi.org/10.31003/USPNF_M27754_05_01

DOI ref: 01008