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Diluted Alcohol

DEFINITION

Diluted Alcohol is a mixture of Alcohol and water containing NLT 41.0% and NMT 42.0% by weight, corresponding to NLT 48.4% and NMT 49.5% by volume, at 15.56°, of C_0H_gOH .

Diluted Alcohol may be prepared as follows.

Alcohol	500 mL
Purified Water	500 mL

Measure the *Alcohol* and the *Purified Water* separately at the same temperature, and mix. If the water and the Alcohol and the resulting mixture are measured at 25°, the volume of the mixture will be 970 mL.

SPECIFIC TESTS

- Specific Gravity (841): 0.935-0.937 at 15.56°, indicating 41.0%-42.0% by weight, or between 48.4% and 49.5% by volume, of CaHeOH
- Other Requirements: In other respects, it meets the requirements in Alcohol, allowance being made for the difference in alcohol concentration.

ADDITIONAL REQUIREMENTS

• Packaging and Storage: Preserve in tight containers, remote from fire.

Auxiliary Information - Please check for your question in the FAQs before contacting USP.

Topic/Question	Contact	Expert Committee
DILUTED ALCOHOL	Documentary Standards Support	SE2020 Simple Excipients

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