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## Dibasic Sodium Phosphate

$\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$	358.14
$\text{Na}_2\text{HPO}_4 \cdot 7\text{H}_2\text{O}$	268.07
$\text{Na}_2\text{HPO}_4 \cdot 2\text{H}_2\text{O}$	177.99
$\text{Na}_2\text{HPO}_4 \cdot \text{H}_2\text{O}$	159.94
$\text{Na}_2\text{HPO}_4$	141.96

Phosphoric acid, disodium salt, dodecahydrate;

Disodium hydrogen phosphate, dodecahydrate CAS RN®: 10039-32-4.

Phosphoric acid, disodium salt, heptahydrate;

Disodium hydrogen phosphate heptahydrate CAS RN®: 7782-85-6.

Phosphoric acid, disodium salt, dihydrate;

Disodium hydrogen phosphate, dihydrate CAS RN®: 10028-24-7.

Phosphoric acid, sodium salt, monohydrate;

Disodium hydrogen phosphate, monohydrate CAS RN®: 118830-14-1.

Phosphoric acid, disodium salt, hydrate;

Disodium hydrogen phosphate hydrate CAS RN®: 10140-65-5.

Anhydrous CAS RN®: 7558-79-4.

### DEFINITION

Dibasic Sodium Phosphate is dried or contains one, two, seven, or twelve molecules of water of hydration. It contains NLT 98.0% and NMT 100.5% of  $\text{Na}_2\text{HPO}_4$ , calculated on the dried basis.

### IDENTIFICATION

- **A.** [IDENTIFICATION TESTS—GENERAL, Sodium\(191\)](#).

**Sample solution:** Equivalent to 1 part of  $\text{Na}_2\text{HPO}_4$  in 30

**Acceptance criteria:** The solution meets the requirements.

- **B.** [IDENTIFICATION TESTS—GENERAL, Phosphate\(191\)](#).

**Sample solution:** Equivalent to 1 part of  $\text{Na}_2\text{HPO}_4$  in 30

**Acceptance criteria:** The solution meets the requirements.

### ASSAY

- **PROCEDURE**

**Sample:** Equivalent to 2.5 g of  $\text{Na}_2\text{HPO}_4$

**Sample solution:** Transfer the *Sample* to a 250-mL beaker. Add 50 mL of water and 40.0 mL of 1 N hydrochloric acid, and stir until dissolved.

**Blank:** Transfer 40.0 mL of 1 N hydrochloric acid to a 250-mL beaker. Add 50 mL of water.

**Analysis:** Titrate the *Blank* with 1 N sodium hydroxide VS, and record the volume of 1 N sodium hydroxide VS consumed. Titrate the excess acid in the *Sample solution* potentiometrically with 1 N sodium hydroxide VS to the inflection point at about pH 4, and record the buret reading. Subtract this buret reading from that of the *Blank*, and designate the volume of 1 N sodium hydroxide VS resulting from this subtraction as A. Continue the titration with 1 N sodium hydroxide VS to the inflection point at about pH 8.8, record the buret reading, and calculate the volume (B) of 1 N sodium hydroxide required in the titration between the two inflection points (pH 4–8.8). Where A is equal to or less than B, each mL of the volume A of 1 N sodium hydroxide is equivalent to 142.0 mg of  $\text{Na}_2\text{HPO}_4$ . Where A is greater than B, each mL of the volume (2B – A) of 1 N sodium hydroxide is equivalent to 142.0 mg of  $\text{Na}_2\text{HPO}_4$ .

**Acceptance criteria:** 98.0%–100.5% on the dried basis

## IMPURITIES

### • INSOLUBLE SUBSTANCES

**Sample solution:** Equivalent to 5.0 g of  $\text{Na}_2\text{HPO}_4$  in 100 mL of hot water

**Analysis:** Filter through a tared filtering crucible, wash the insoluble residue with hot water, and dry at 105° for 2 h.

**Acceptance criteria:** NMT 20 mg (NMT 0.4%)

### • CHLORIDE AND SULFATE, [Chloride\(221\)](#)

**Sample:** Equivalent to 0.5 g of  $\text{Na}_2\text{HPO}_4$

**Acceptance criteria:** Shows no more chloride than corresponds to 0.42 mL of 0.020 N hydrochloric acid (NMT 0.06%)

### • CHLORIDE AND SULFATE, [Sulfate\(221\)](#)

**Sample:** Equivalent to 0.1 g of  $\text{Na}_2\text{HPO}_4$

**Acceptance criteria:** Shows no more sulfate than corresponds to 0.2 mL of 0.020 N sulfuric acid (NMT 0.2%)

### Change to read:

### • [ARSENIC \(211\)](#), [Procedures, Procedure 1](#) ▲ (CN 1-JUN-2023)

**Test preparation:** Equivalent to 187.5 mg of  $\text{Na}_2\text{HPO}_4$  in 35 mL of water

**Acceptance criteria:** NMT 16 ppm

## SPECIFIC TESTS

### • [LOSS ON DRYING \(731\)](#): Dry a sample at 130° to constant weight.

**Acceptance criteria:** See [Table 1](#).

Table 1

Hydrate Form	Acceptance Criteria
Dried	NMT 5.0%
Monohydrate	10.3%–12.0%
Dihydrate	18.5%–21.5%
Heptahydrate	43.0%–50.0%
Dodecahydrate	55.0%–64.0%

## ADDITIONAL REQUIREMENTS

### • PACKAGING AND STORAGE: Preserve in tight containers.

### • LABELING: Label it to indicate whether it is dried or is the monohydrate, the dihydrate, the heptahydrate, or the dodecahydrate.

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
DIBASIC SODIUM PHOSPHATE	<a href="#">Documentary Standards Support</a>	SM32020 Small Molecules 3
REFERENCE STANDARD SUPPORT	RS Technical Services <a href="mailto:RSTECH@usp.org">RSTECH@usp.org</a>	SM32020 Small Molecules 3

**Chromatographic Database Information:** [Chromatographic Database](#)

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