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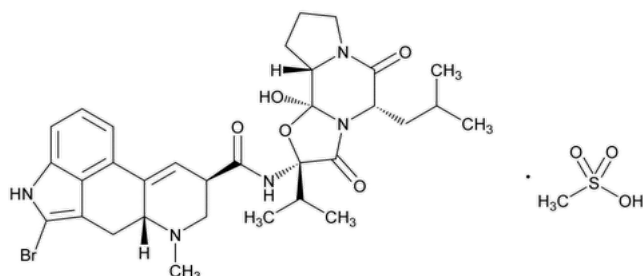
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Bromocriptine Mesylate


 $C_{32}H_{40}BrN_5O_5 \cdot CH_4SO_3$ 750.70

Ergotaman-3',6',18-trione, 2-bromo-12'-hydroxy-2'-(1-methylethyl)-5'-(2-methylpropyl)-, monomethanesulfonate (salt), (5'α)-;

2-Bromoergocryptine monomethanesulfonate (salt) CAS RN®: 22260-51-1; UNII: FFP983J30D.

DEFINITION

Bromocriptine Mesylate contains NLT 98.0% and NMT 102.0% of $C_{32}H_{40}BrN_5O_5 \cdot CH_4SO_3$, calculated on the dried basis.

IDENTIFICATION

Change to read:

- A. [▲ SPECTROSCOPIC IDENTIFICATION TESTS \(197\), Infrared Spectroscopy: 197M](#)▲ (CN 1-MAY-2020): Undried

Change to read:

- B. [▲ SPECTROSCOPIC IDENTIFICATION TESTS \(197\), Ultraviolet-Visible Spectroscopy: 197U](#)▲ (CN 1-MAY-2020)

Sample solution: 50 µg/mL in 0.1 M methanolic methanesulfonic acid**Acceptance criteria:** Meets the requirements

ASSAY

PROCEDURE

Sample solution: 600 mg of Bromocriptine Mesylate**Analysis:** Dissolve with 80 mL of a mixture of acetic anhydride and glacial acetic acid (7:1). Titrate with 0.1 N perchloric acid VS. Perform a blank determination, and make any necessary correction (see [Titrimetry \(541\)](#)). Each mL of 0.1 N perchloric acid is equivalent to 75.07 mg of $C_{32}H_{40}BrN_5O_5 \cdot CH_4SO_3$.**Acceptance criteria:** 98.0%–102.0% on the dried basis

IMPURITIES

INORGANIC IMPURITIES

- [RESIDUE ON IGNITION \(281\)](#): NMT 0.1%

ORGANIC IMPURITIES

PROCEDURE 1: LIMIT OF METHANESULFONIC ACID CONTENT

Sample solution: 400 mg of Bromocriptine Mesylate**Analysis:** Dissolve with 70 mL of methanol. Titrate under nitrogen with 0.1 N methanolic potassium hydroxide VS. Perform a blank determination, and make any necessary correction (see [Titrimetry \(541\)](#)). Each mL of 0.1 N methanolic potassium hydroxide is equivalent to 9.61 mg of CH_3SO_3H .**Acceptance criteria:** NLT 12.5% and NMT 13.4% of CH_3SO_3H on the dried basis

PROCEDURE 2

Solution A: 0.1 N citric acid solution. Adjust with hydrochloric acid to a pH of 2.0.**Diluent:** Methanol and *Solution A* (1:1)**Solution B:** Acetonitrile and 0.01 M phosphate buffer, pH 7.0 (2:3)**Solution C:** Acetonitrile and 0.01 M phosphate buffer, pH 7.0 (3:2)

Mobile phase: See the gradient table below.

Time (min)	Solution B (%)	Solution C (%)
0	100	0
18	100	0
30	0	100
40	0	100
41	100	0

System suitability solution: 2.0 mg/mL each of α -ergocryptine and Bromocriptine Mesylate in *Diluent*

Standard stock solution: 46 μ g/mL of [USP Bromocriptine Mesylate RS](#) in methanol and *Solution A* (1:1). [NOTE—Dissolve in 50% of the flask volume of methanol and dilute with *Solution A* to volume.]

Standard solution: 4.6 μ g/mL of [USP Bromocriptine Mesylate RS](#) in *Diluent* from the *Standard stock solution*

Sample solution: 4.6 mg/mL of Bromocriptine Mesylate in methanol and *Solution A* (1:1). [NOTE—Dissolve in 50% of the flask volume of methanol and dilute with *Solution A* to volume.]

Chromatographic system

(See [Chromatography \(621\)](#), *System Suitability*.)

Mode: LC

Detector: UV 300 nm

Column: 4.6-mm $\times$ 15-cm; 3- μ m packing L1

Flow rate: 2 mL/min

Injection size: 20 μ L

System suitability

Samples: *System suitability solution* and *Standard solution*

[NOTE—The relative retention times for α -ergocryptine and bromocriptine mesylate are 0.46 and 1.0, respectively.]

Suitability requirements

Resolution: NLT 15 between α -ergocryptine and bromocriptine mesylate, *System suitability solution*

Tailing factor: NMT 1.5, *System suitability solution*

Relative standard deviation: NMT 10.0%, *Standard solution*

Analysis

Samples: *Standard solution* and *Sample solution*

[NOTE—The relative retention times for bromocriptine and bromocriptinine are 1.0 and 1.7, respectively.]

Calculate the percentage of each impurity in the portion of Bromocriptine Mesylate taken:

$$\text{Result} = (r_U/r_S) \times (C_S/C_U) \times (1/F) \times 100$$

r_U = peak response of each impurity from the *Sample solution*

r_S = peak response of bromocriptine from the *Standard solution*

C_S = concentration of [USP Bromocriptine Mesylate RS](#) in the *Standard solution* (mg/mL)

C_U = concentration of Bromocriptine Mesylate in the *Sample solution* (mg/mL)

F = relative response factor equal to 1.4 for any peak eluting at a relative retention time of about 0.9 or less, and equal to 1.0 for all other peaks

Acceptance criteria

Individual impurities: NMT 0.4% of bromocriptinine is found; NMT 0.1% of any individual impurity is found.

Total impurities: NMT 1.0%

SPECIFIC TESTS

• [COLOR OF SOLUTION \(631\)](#)

Matching solutions: Prepare three solutions, A, B, and C, containing, respectively, the following parts of cobaltous chloride CS, ferric chloride CS, cupric sulfate CS, and dilute hydrochloric acid (1 in 40).

A: 3.0:3.0:2.4:31.6

B: 1.0:2.4:0.4:36.2

C: 0.6:2.4:0:37.0

Sample solution: 10 mg/mL of Bromocriptine Mesylate in methanol

Analysis: Compare the *Sample solution* with 10-mL portions of the *Matching solutions* in suitable matched tubes.

Acceptance criteria: The solution is clear and not darker in color than *Matching solutions A, B, and C*.

- [OPTICAL ROTATION, Specific Rotation \(781S\)](#).

Sample solution: 10 mg/mL, in a mixture of methylene chloride and methanol (1:1)

Acceptance criteria: +95° to +105°

- **LOSS ON DRYING**

(See [Thermal Analysis \(891\)](#).)

Analysis: Determine the percentage of volatile substances by thermogravimetric analysis using 10 mg of Bromocriptine Mesylate. Heat the specimen under test at the rate of 10°/min in an atmosphere of nitrogen at a flow rate of 45 mL/min. Record the thermogram from ambient temperature to 160°.

Acceptance criteria: It loses NMT 4.0% of its weight.

ADDITIONAL REQUIREMENTS

- **PACKAGING AND STORAGE:** Preserve in tight, light-resistant containers, in a cold place.

- **USP REFERENCE STANDARDS (11).**

[USP Bromocriptine Mesylate RS](#)

Auxiliary Information - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
BROMOCRIPTINE MESYLATE	Documentary Standards Support	SM52020 Small Molecules 5

Chromatographic Database Information: [Chromatographic Database](#)

Most Recently Appeared In:

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