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# Aluminum Chloride

$\text{AlCl}_3 \cdot 6\text{H}_2\text{O}$  241.43  
 $\text{AlCl}_3$  133.34

Aluminum chloride, hexahydrate;

Aluminum chloride hexahydrate CAS RN®: 7784-13-6; UNII: 3CYT62D3GA.

Anhydrous CAS RN®: 7446-70-0; UNII: LIF1N9568Y.

## DEFINITION

Aluminum Chloride contains NLT 95.0% and NMT 102.0% of aluminum chloride ( $\text{AlCl}_3$ ), calculated on the anhydrous basis.

## IDENTIFICATION

• **A. IDENTIFICATION TESTS—GENERAL, [Aluminum\(191\)](#) and [Chloride\(191\)](#).**

**Sample solution:** 100 mg/mL

**Acceptance criteria:** Meets the requirements

## ASSAY

### • PROCEDURE

**Edetate disodium titrant:** Prepare and standardize as directed in *Reagents, Volumetric Solutions, Edetate Disodium, Twentieth-Molar (0.05 M)*.

**Sample solution:** 20 mg/mL of aluminum chloride in water

#### Titrimetric system

**Mode:** Back-titration

**Titrant:** 0.05 M zinc sulfate VS

**Endpoint detection:** Visual

**Analysis:** Transfer 10.0 mL of the *Sample solution* into a 250-mL beaker, and add, in the order named and with continuous stirring, 25.0 mL of *Edetate disodium titrant* and 20 mL of acetic acid–ammonium acetate buffer TS, and boil gently for 5 min. Cool, and add 50 mL of alcohol and 2 mL of dithizone TS. Titrate the excess edetate disodium with *Titrant* to a bright rose-pink color. Perform a blank determination, substituting 10 mL of water for the *Sample solution*, and make any necessary correction. Each mL of *Edetate disodium titrant* consumed is equivalent to 6.667 mg of aluminum chloride ( $\text{AlCl}_3$ ).

**Acceptance criteria:** 95.0%–102.0% on the anhydrous basis

## IMPURITIES

### • LIMIT OF SULFATE

**Sample solution:** 10 mg/mL

**Analysis:** Add 0.2 mL of barium chloride TS to 10 mL of the *Sample solution*.

**Acceptance criteria:** No turbidity is produced within 1 min.

### • [IRON \(241\)](#)

**Sample:** 1.0 g

**Analysis:** Dissolve the *Sample* in 45 mL of water, and add 2 mL of hydrochloric acid.

**Acceptance criteria:** NMT 10 ppm

### • LIMIT OF ALKALIES AND ALKALINE EARTHS

**Sample:** 1.0 g

**Analysis:** To a boiling solution of the *Sample* in 150 mL of water add a few drops of methyl red TS, then add 6 N ammonium hydroxide until the color of the solution just changes to a distinct yellow. Add hot water to restore the volume to 150 mL, and filter while hot. Evaporate 75 mL of the filtrate to dryness, and ignite to a constant weight.

**Acceptance criteria:** 0.5%; the weight of the residue is NMT 2.5 mg.

## SPECIFIC TESTS

• **[WATER DETERMINATION, Method I \(921\)](#):** 42.0%–48.0%

## ADDITIONAL REQUIREMENTS

• **PACKAGING AND STORAGE:** Preserve in tight containers.

**Auxiliary Information** - Please [check for your question in the FAQs](#) before contacting USP.

Topic/Question	Contact	Expert Committee
ALUMINUM CHLORIDE	<a href="#">Documentary Standards Support</a>	SM32020 Small Molecules 3

**Chromatographic Database Information:** [Chromatographic Database](#)

**Most Recently Appeared In:**

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